

A 'DAY IN THE LIFE' OF THE HUDSON RIVER ESTUARY October 20, 2015

http://www.ldeo.columbia.edu/edu/k12/snapshotday/

PLEASE INCLUDE THIS COVER SHEET WITH YOUR SUBMITTAL

- These recording sheets contain every testing option but you can chose to do any combination with your samplers and be part of Day In The Life.
- More activities to support your field experience are available at the website link above.
- Please submit your results to Margie Turrin (845-365-8179 (fax) or e-mail mkt@ldeo.columbia.edu) within 24-48 hours of collection! Questions? 845-365-8494. PLEASE include this cover sheet with any data sheets your submit. THANK YOU!

RECORDING SHEET I - BACKGROUND INFORMATION. 1. Site contact person Organization_____ City____State__Zip____ phone fax email_ 2. School/group name District Name of teacher/group leader Street____ City_____State____Zip____ phone_____fax___email____ Number of school students Number of Adults grade level/High School course 3. Where are you sampling. Please be as specific as possible. (Example: swimming beach, Kingston Point, City of Kingston, Ulster County.) 4. Using the map included with your packet, give your location along the Hudson estuary in river miles. (The Battery at the southern tip of Manhattan is River Mile 0; the Federal Dam at Troy is River Mile 153.) River mile If you have a way to determine the latitude and longitude of your site, enter that here. GPS Latitude Longitude

PLEASE BE SURE TO RECORD TIMES & UNITS OF MEASURE SO THAT RESULTS CAN BE COMPARED AT DIFFERENT SPOTS ON THE HUDSON

Name	DITL	2015 Observing	Surroundings	Location
Let's look at your sampling	ng site.	At this station we wil	I describe the site	and any plants in and
around your collection ar	rea. Incl	lude plant materials ir	n the water, as well	as along the water's edge.



1. Sketch your sampling site. Be sure to include a compass rose showing N/S/E/W and locate the River on your sketch, and note your sampling site.

2. Land type around your sample site? Check all	PierGrassy
that apply.	ForestedParking Lot
3. <u>Surrounding land uses as percentage</u> of 100 (for example 50% is half the usage)	Urban/residentialForested Industrial/commercialBeach
(101 example con is harf the asage)	Other
4. Describe the shoreline - check percentage	BeachCovered in vegetation
below and then all that apply in next column:	Banks alteredRipRap (Large rocks)
	Wood BulkheadConcrete Bulkhead
Sandy MuddyRocky	Pipe entering the water
5. Describe the water area at the sampling site	DepthBottom sandy
	Bottom muddyBottom rocky
	Bottom weedy
6. Describe the water itself	CalmChoppy
7. Plants in the water (water chestnut, water	% Plant
celery etc.) that you have identified & percent	% Plant
of total area covered. IF NONE please check	% Plant
None.	% Plant
	No Plants in the water area

Name

Time:

Name	

DITL 2015 Physical Conditions Data Location_____

(weather, tides, currents)

1.	location. Location	n Name	ips along the Hudson, so it is important to note our exact				
	Using your Hudson River Estuary map, give your location in <i>river miles</i> : and if possible						
	GPS Latitude:	GPS	Longitude:				
	2. Tides : Tides	cause the water of the Hudson	Start time:*				
	River to rise o	and fall due to the gravitational	Check time: Water height in cm				
		n and the moon. Tides can be	Check time: Water height in cm				
	•	r time with a Tide Meter Stick , or	Check time: Water height in cm				
		er to see if the water is rising,	Check time: Water height in cm				
		ying the same. First record the	Check time: Water height in cm				
	-	eck the water level using your	Check time: Water height in cm				
		ck or tape. Check the stick again	-				
	-	ery 15 to 30 minutes) and record.	*if on a dock measure down from the dock				
		rrents record the direction of	Time: Circle: ebb - flood - still*				
	water moveme	ent. A current moving downriver is	Cm/30secsCm/secKnots				
		current moving upriver it is called					
		there is no current it is <i>still</i> . Toss	Time: Circle: ebb - flood - still*				
	an orange or o	a solid stick as far as you can out	Cm/30secsCm/secKnots				
	_	and watch to see which way it					
	moves. Recor	d: Ebb or Flood or Still. Time its	Time: Circle: ebb - flood - still*				
	movement for	30 secs and record, then divide	Cm/30secsCm/secKnots				
	by 30 to get p	per cm/sec of movement.					
	To determin	e knots: measure distance orange	Time: Circle: ebb - flood - still				
	or stick trave	els in 30 secs. Divide by 30 for	Cm/30secsCm/secKnots				
	cm/sec and th	nen divide by 51.4 for knots. The	* Note if anything about the shoreline could cause the				
	formula for k	nots is (cm/sec)/51.4	current near shore to flow in a different direction				
			than the current in the middle of the Hudson?				
	4. Air Tempero	ature: How to convert:	Time:Air temperature:°F°C				
	°C = 0.556	X (°F - 32°)	Time:Air temperature:°F°C				
	°F = (1.8 X	°C) + 32°	Time:Air temperature:°F°C				
			//////////////////////////////////////				
	_	If you use the Beaufort chart	Time:Beaufort # Wind Meter: Units				
		ort force #. If you use a wind	Wind Direction (comes from)				
	meter record	number registered AND units.	Wind Bill ection (comes from)				
		Select from the scale provided a	Time:				
	percentage of	f cloud cover.	clearpartly cloudymostly cloudy,overcast				
			(<25%) (26-50%) (51-75%) (>75%)				
	7. Rain (Precipit	ation) Today & Weather for the	Time:Rain If checked note how steadily it				
	•	Rain can affect our readings and					
	•	ne changes in temperature over a	rained				
		We record weather today and for	Briefly describe the weather for the last 3 days: Rain,				
	the last 3 day	•	·				
			wind, unusual temperatures?				
		·	·				

Name	Location	Time
2015 SALIN	VITY Recording Sheet: When we me	asure salinity we are measuring the
source of salt i small amounts o * Expected Hu Put a checkmar	present in water. Much of this salt is sodiu in the Hudson is seawater (~35,000 ppm) po of salt in the fresh water entering the rive dson Range: ~40 ppm in the freshwater sec rk in the box next to the measuring method test and then record your results below.	ushing in from the ocean. There are only r from the eroding rocks and road salt. tion to ~29,000 ppm in harbor*
 TITRATOR Low Range Tabs 	STRIPS measure chloride by color change S High Range Tabs On strip's E conversion table and RECORD: Units	scale, white color ends at
	NT TEST KITS usually measure chloride us ample drop by drop.	ing color change as a liquid chemical is
How many drops	s were needed for the sample to change color	? drops
Number of drop	os times conversion factor (from instruction	s) equals chloride concentration.
X_	= mg/L Cl	
	TERS measure water's density (its specific ses, density increases, and the object floats	
If u <u>sing a</u> plast	ic hydrometer with a pointer, record salinity	here: parts per thousand (ppt)
If u <u>sing a glass</u> ° <i>C</i>	hydrometer floating in a water sample: Reco	rd the temperature of the water sample
breaks the wate	ecific gravity (to the fourth decimal place) fr er's surface. Read at water level, not at the t	op of the meniscus
Record salinity	from the specific gravity conversion table: _	parts per thousand (ppt)
	NETERS measure how light is bent—refract which in turn varies with salinity (density inc	
Read salinity wh	nere the shadowline crosses the display scale	: parts per thousand (ppt)
	asure how well water conducts electricity (brity, salinity, or chloride concentration; be s	•
Reading	Units of measurement	

pH - Expected Range - Most fish prefer 6.5 to 8.5 - pH measures how acidic or basic (a s measured on a scale from 0 to 14. Neutral is 7.0, Acidic is lower than 7.0 and Basic is higher are NO UNITS used with pH. Circle equipment used for the test: Test Strips	
s measured on a scale from 0 to 14. Neutral is 7.0, Acidic is lower than 7.0 and Basic is higher re NO UNITS used with pH. Circle equipment used for the test: Test Strips	
Time: Reading 1: Reading 2: Reading 3: Reading 3: Time: Reading 1: Reading 2: Reading 3: Reading 3: Time: Reading 1: Reading 2: Reading 3: Reading 3: Time: Reading 1: Reading 2: Reading 3: Reading 3: Time: Reading 1: Reading 2: Reading 3: Reading 3: Reading 3: Reading 2: Reading 3: Reading 3: Reading 3: Reading 3: Reading 3: Reading 3: Reading 4: Reading 5: Reading 5: Reading 5: Reading 6: Readi	
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Time:	
**Salinity - Expected Range - ~40 ppm in freshwater section up to 29,000 ppm measure of the amount of salt present in the water (we are measuring the chloride). Or measured in 'Parts per thousand' (PPT). Fresher water with smaller amounts in 'Parts per mg/l. (Note: There are other measures of salinity: 'Practical salinity units' (PSU), Conductivity as mS/ci uS/cm (Microsiemens) but we focus on ppt and ppm) Circle equipment used for the test (what each measures is listed below each equipment Drop count test kit Quantab strips meter refractometer hydr (chloride) (chloride) (PPT or mS/cm) (PPT) (PPT Time: Reading 1: Reading 2: Reading 3: (note correct Time: Reading 1: Reading 2: Reading 3: (note correct Time: Reading 1: Reading 2: Reading 3: (note correct Time: Reading 1: Reading 2: Reading 3: (note correct *(NOTE THERE IS AN INDIVIDUAL CALCULATION SHEET FOR SALINITY) * Water temperature Expected High Temperature in October would by < 25 Record water temperature in degrees Celsius or degrees Fahrenheit. TO convert between to C = 0.556 X ("F - 32") "F = (1.8 X "C) + 32" * Time: water depth (feet): Reading 1: °C °F Reading 2: °C °F Average: °C * Time: water depth (in feet): Reading 1: °C °F Reading 2: °C Average: °C °C °F Reading 1: °C °F Reading 2: °C Average: °C	
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Reading 1: °C °F °F Reading 2: °C Average: °C Observations	_ ° F
Reading 1: °C °F °F Reading 2: °C Average: °C Observations	
Observations	
	_ ° F
Describe your water collection site -	
direct sunshadeWater covered with plantsWater very still	n
What else should we know about your sampling?	

Dissolved oxygen (DO)

Healthy Expected Range 5.0-11.0~mg/L

Name	DITL 2015	Chemical Des	scription Location	i
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The amount of dissolved oxygen in the water is one of the most important factors in telling how healthy that ecosystem is. Many variables affect DO, including temperature, time of day, presence of plants, and wind conditions. DO measurements are given in mg/l and as percent saturation. 100% saturation means that the water cannot hold any more oxygen at that temperature. If more oxygen is added (such as by a high wind or a waterfall) the oxygen will go from the water into the air. Circle equipment used for the test:

meter drop count kit other

For test temperature use average from water temperature (#3) above. For % saturation calculation use chart on bottom of page.

* Time: ______Water temperature in °C _____ DO (mg/l) or PPM: ____% saturation _____

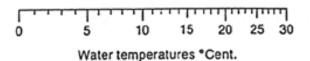
* Time: _____Water temperature in °C _____ DO (mg/l) or PPM: ____% saturation _____

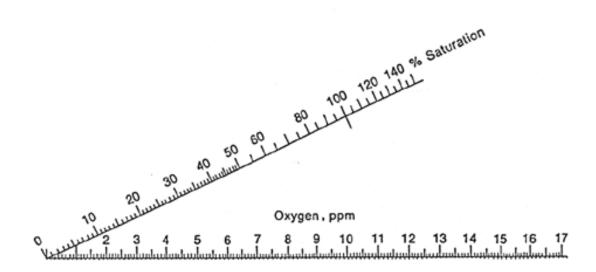
* Time: ______Water temperature in °C _____ DO (mg/l) or PPM: ____% saturation ______

5. % Saturation of Dissolved Oxygen (DO) Healthy Expected Range 90% or above

Source of chart: http://waterontheweb.org/under/waterquality/oxygen.html

For a quick and easy determination of the percent saturation value for dissolved oxygen at a given temperature, use the saturation chart below. Locate your DO reading on the bottom scale (ppm equals mg/L). Locate the temperature of the water in degrees C on the top scale. Draw a straight line between the temperature and DO. The % saturation is the value where the line intercepts the saturation scale.





Na	me:	2015 Turbio	lity	Location:	_
1.	water than turbid. Tur dead plants - be careful	it can in turbid wate bidity can be caused , salt, sand and mud.	er. Estuaries lik d by phytoplank Measure the to g water for a si	an penetrate farther in clear we the Hudson River are naturally ton and zoo planktons, bits of urbidity at your site on the river te tube NOT to step in the wate ur sample.	
	Time:		<u>Circle eq</u>	uipment used	
1)5	iecchi disk (d	cm) 2) Long Sigh	t tube (cm)	3) Short Site Tube (JTU)	
4)7	Turbidimeter	(NTU)			
	Reading	Reading 2	Reading	3 Average	
(M		t you RECORD the c eet, cm, meters, JT		the piece of equipment that you	
Is	the water re	eally turbid? How w	Observations ould you describ	pe it in words?	

NOTE in 2015 we will not be collecting chlorophyll samples.

···· -	LENGT	H OF NE	<u> </u>			
sh Species:		# of	individuals:	Siz	e of largest	(unit)
acroinvertebrat pe (blue, mud,	tes (For crabs includ Asian etc.) & sex (A # ## ####_	e M/F) — —				
acroinvertebrat be (blue, mud,	tes (For crabs includ Asian etc.) & sex (A _## ##	e M/F) — —				
acroinvertebrat be (blue, mud,	tes (For crabs includ Asian etc.) & sex (A # ## ####_	e M/F) — —				
acroinvertebrat	tes (For crabs includ Asian etc.) & sex (A ## # # #	e A/F) — —				

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DAY in the LIFE PUSH CORE SEDIMENT LOG

GRAB ID#	Site I	Name	DATE		FORM COMPLETED BY:			
					GROUP#			
TIME	LATI	TUDE	LONGITUDE		WATER DEPTH	LOCATION		
	Yes	No			Descriptors - Please note	additional observations		
H₂S smell				,	H ₂ S smells of rotten eggs, suggesting anaerobic ba			
Oil				Oil creates a slight smell, a slickness and a				
Oxidized top*					*oxidation (reaction with oxygen) creates a distinctly lighter colored layer of sediment.			
					estimate dimensions of oxided layer, etc. and draw bel			
	Absent	Rare	Common	Abundant	Additional Comments			
Clay					very fine material - grey co	olor & rich dense feel		
Mud					smooth feel between finge	ers - brown color		
Sand					gritty feeling between fing	ers		
Gravel					pea sized pieces of stone			
Pebbles					pieces of stone larger than	n pea		
Leaves								
Wood								
Shells Oysters (dead/alive?)								
Freshwater mussels (except zebra)								
Zebra mussels								
macroinvertebrates								
Brick								
Coal								
Slag					industrial byproduct - chu	nky look, light, air filled		
Living vegetation:								
Length of Core:					Length of Oxidized core top (if present):			
If Bagged - Number On	Core Collec	tion Bag						
Sketch of your core below with measurements for each section & total core (be sure to label the top and bottom):								
			<botto< td=""><td>м</td><td>TOP></td><td></td></botto<>	м	TOP>			
CBOTTOWI TOF>								